

Synergistic effect of hydroxyapatite nanopowder's high crystallinity and non-ordered particles' boundary regions on low-temperature sintering

Miodrag J. Lukić, Ljiljana Veselinović, Smilja Marković and Dragan Uskoković

Centre for Fine Particles Processing and Nanotechnologies, Institute of the Technical Sciences of the SASA, Belgrade, Serbia

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Introduction

- * Fabrication of fully dense nanostructured hydroxyapatite is one way to get **reliable bone tissue implant for large defect repairs**.
- * It is expected that nanopowders would enhance sintering behavior!
- * Generally, what should be fulfilled?
 - average particle size ≤ 100 nm
 - uniform particle size distribution
 - specific surface area ≥ 50 m²/g
 - spherical particles morphology

However, results are quite versatile!

Table I. Summary of the current results of the pressureless sintering of the hydroxyapatite.

Synthesis method	Morphology	Size length/width (nm)	SSA (m ² /g)	Xc (%)	Sintering method	Sintering temp. (°C)	Final relative Density (%)	Average grain size (nm)
precipitation	needle-like	100	58	low	CS	1250	97	10 ³
precipitation	spherical	600	48	-	CS	1200	76.8	-
precipitation + milling	nanorods	50/15	120	-	CS	850	~99	83
neutralization	elongated	<100	15	80	CS	1215	92.3	10 ³
reverse micelle templating	rod-like	~100	90	-	CS	1200	98.7	>10 ³
hydrothermal	spherical	70/20	39	-	CS	1200	97.5	-
precipitation	needle-like	150/40	-	-	TSS	1050	99	193
commercial	semirods	93/24	-	-	TSS	900	98.8	190
hydrothermal	slightly elongated	65/25	49.5	72	TSS	900	99.2	75

Where should we look for the potential explanations of very distinct sintering behavior?

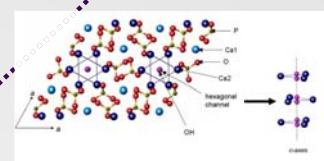
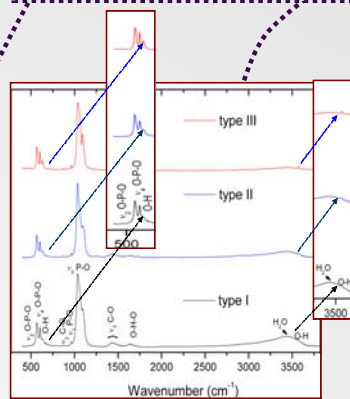
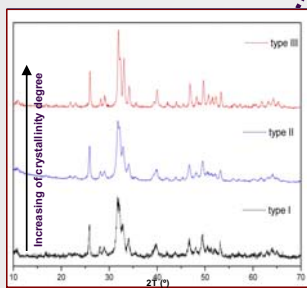
Results

XRD analysis: all systems are phase pure hydroxyapatite.

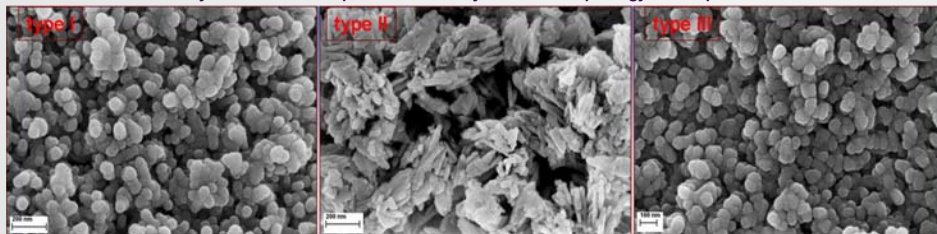
FTIR spectroscopy confirmed chemical purity, but also indicated difference between hydroxyl group vibration intensity implying different level of crystal ordering.

Experimental part

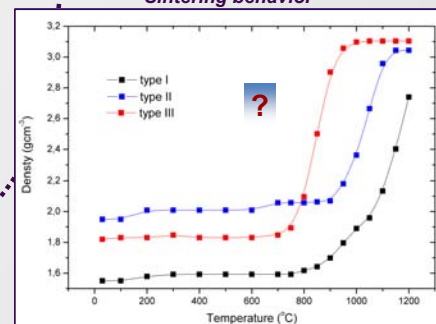
In this study, three types of hydroxyapatite nanopowder were prepared. The precursor chemicals used were Ca(NO₃)₂ and H₃PO₄ for hydroxyapatite types I and III, while for type II as a phosphate source NH₄H₂PO₄ is taken. Chemical precipitation was the basic reaction whereas hydrothermal processing was additionally employed in the case of type III. Materials characterization was performed by XRD, FTIR spectroscopy, scanning and transmission electron microscopy. Moreover, their sintering behavior was investigated by heating microscopy.



FE SEM analysis revealed nanopowders with very different morphology: from spheres to rods!

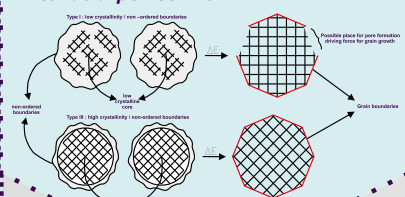


Sintering behavior

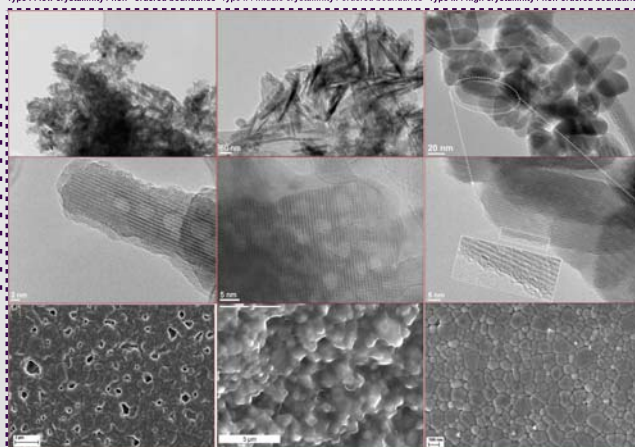


Discussion

...if the situation is considered from the interfacial point of view....



Type I: low crystallinity / non-ordered boundaries Type II: middle crystallinity / ordered boundaries Type III: high crystallinity / non-ordered boundaries



Conclusions

Hydroxyapatite nanopowders could be easily produced; however, they sinter quite differently!!!

We considered sintering behavior of three hydroxyapatite nanopowders from the point of view of particles boundary structure and overall crystallinity. Possibility to obtain fully dense nanostructured bioceramics is discussed.

Here is shown that highly crystalline particles with non-ordered boundaries are favorable for enhanced sintering since thermal energy supplied is mainly used for matter diffusion along grain boundaries which volume fraction is significant when deals with nanopowders.

Further grain boundary engineering should be done to prove this hypothesis...

Acknowledgements

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References

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